

**application of co emission m<sup>57</sup>co mossbauer spectroscopy to ...** - abstract emission m<sup>57</sup>co mossbauer spectroscopy with the <sup>57</sup>co isotope was used to study very dilute rapidly frozen aqueous solutions of cobalt(ii) complexes with low-molecular-weight **an analysis of quadrupole splitting of the mossbauer ...** - mossbauer spectra of human liver ferritin and some pharmaceutically important iron-dextran<sup>57</sup> complexes as ferritin models were measured at 87 k (frozen solutions) and at both 87 and 295 k (lyophilized forms). **m<sup>57</sup>co mossbauer spectroscopic study of frozen solutions of ...** - mossbauer spectroscopic study of frozen solutions of <sup>57</sup>co using a specially designed cryostat for time differential coincidence emission mossbauer spectroscopy of <sup>237</sup>np masakatsu saeki a, masami nakada a, nobuyuki m. masaki a, zenko yoshida a, ... **dynamics of hemoglobin investigated by m<sup>57</sup>co mossbauer spectroscopy** - spin resonance and mossbauer spectroscopy that in frozen solutions above 210k there is a dynamic equilibrium with a new low-spin species which we identified as a complex where both the proximal and distal histidines are bound to iron and the **mossbauer spectroscopy of organometallic compounds in ...** - crystalline matrices: mossbauer spectroscopy of frozen aqueous solutions and nonaqueous solutions such as benzene and acetonitrile is useful in the investigation of (1) ions "trapped" in ice, **supplementary information corrected - rsc** - supplementary information to prove the formation of fe(oh)<sub>2</sub> as an intermediate in the synthesis of ferroxhyte nanoparticles a mossbauer spectroscopy of frozen aqueous solutions (fas) were used. **changes apparent ph on freezing aqueous buffersolutions ...** - includes e.p.r.\* spectroscopy, mossbauer spectroscopy, optical spectroscopy and measurements of magnetic susceptibility and magnetic circular dichroism. it is well known that freezing is likely to modify systems under study in some way, and that therefore direct correlation between results obtained at ambient temperatures in solution and results obtained in the frozen state may not always be possible. **studies on the reactions of ferric iron with ascorbic acid ...** - table i. <sup>57</sup>fe mossbauer spectroscopic data on solids and frozen solution materials prepared from fecl<sub>3</sub> and ascorbic acid phase 6 (mm s<sup>-1</sup>) a (mm s<sup>-1</sup>) r s-q absorption (%) frozen solutions **a 19f, 119sn nuclear magnetic resonance and 119sn ...** - concentration of the solutions used for mossbauer spectroscopy was temperature and the frozen solution spectrum recorded again usually -1 m in sn except for the potassium salts where a concentration - the spectrum showed that some augmentation of the sn(iv)

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